

NEET- 2020- 45 Days Crash Course



Date : 17th August 2020



Chapter Name : Rerspiration in Plants



CELLULAR RESPIRATION , TYPES
STEPS – GLYCOLYSIS , KREBS CYCLE , ETS
AMPHIBOLIC PATHWAY
ETS

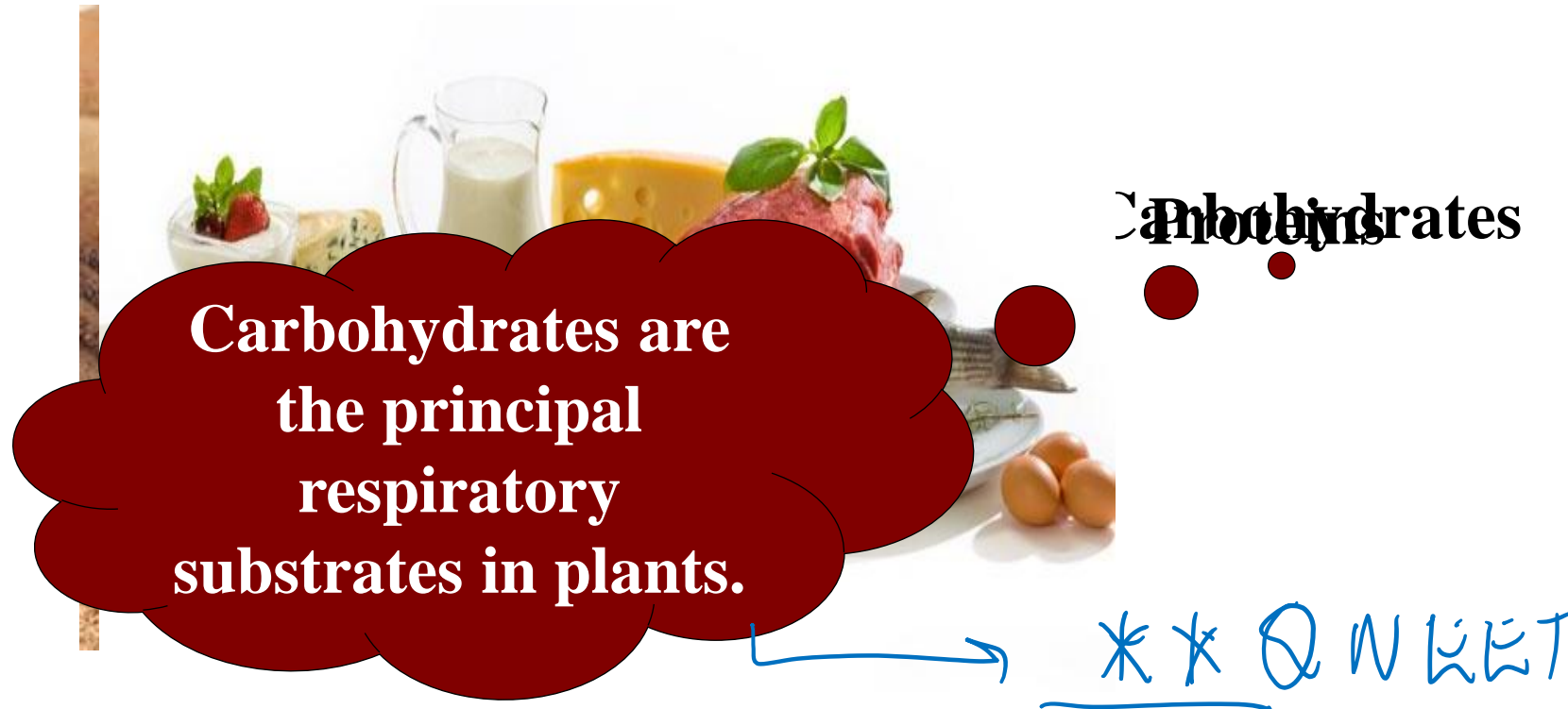
❖ **Cellular respiration** or the mechanism of breakdown of food material within the cell to release energy and the trapping of this energy for the synthesis of ATP.



❖ Photosynthesis, of course, takes place within the chloroplasts (in the eukaryotes), whereas the breakdown of complex molecules to yield energy takes place in the cytoplasm and in the mitochondria (also only in eukaryotes).

❖ The breaking of the C-C bonds of complex compounds through oxidation within the cells, leading to release of considerable amount of energy, is called **respiration**.

- ❖ The compounds that are oxidized during this process are known as **respiratory substrates**.
- ❖ Some of the important respiratory substrates which yield energy in respiration are...



Fats



Organic acids

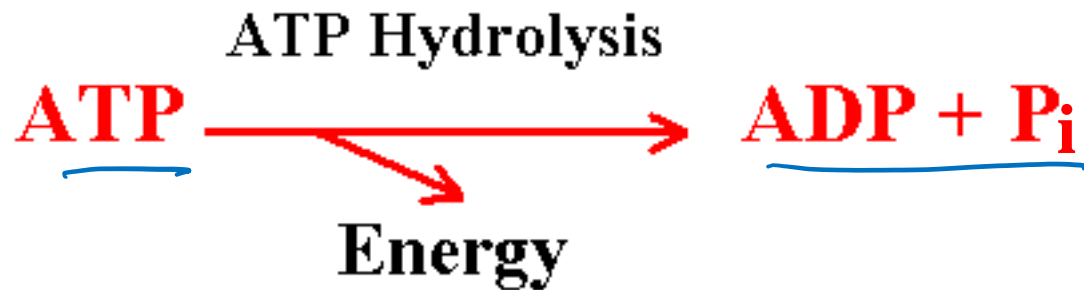


- ❖ Carbohydrates are oxidized to release energy, but proteins, fats, organic acids can be used as respiratory substrates in some plants.

- ❖ Energy released by oxidation in respiration is not used directly but is used to synthesize ATP.

ATP acts as the energy currency of the cell.

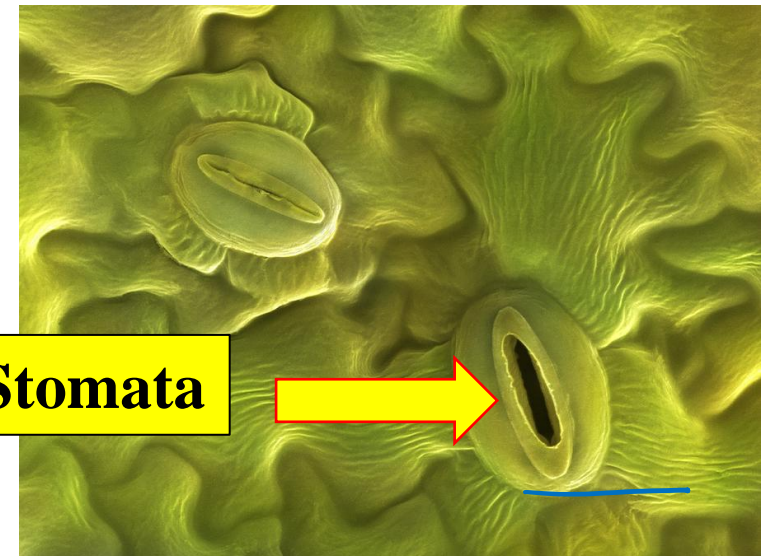
ATP is broken down whenever energy needs to be utilized.



- ❖ Plants unlike animals do not have any specialized organs for gaseous exchange but they have stomata and lenticels for this purpose.



woolly skin



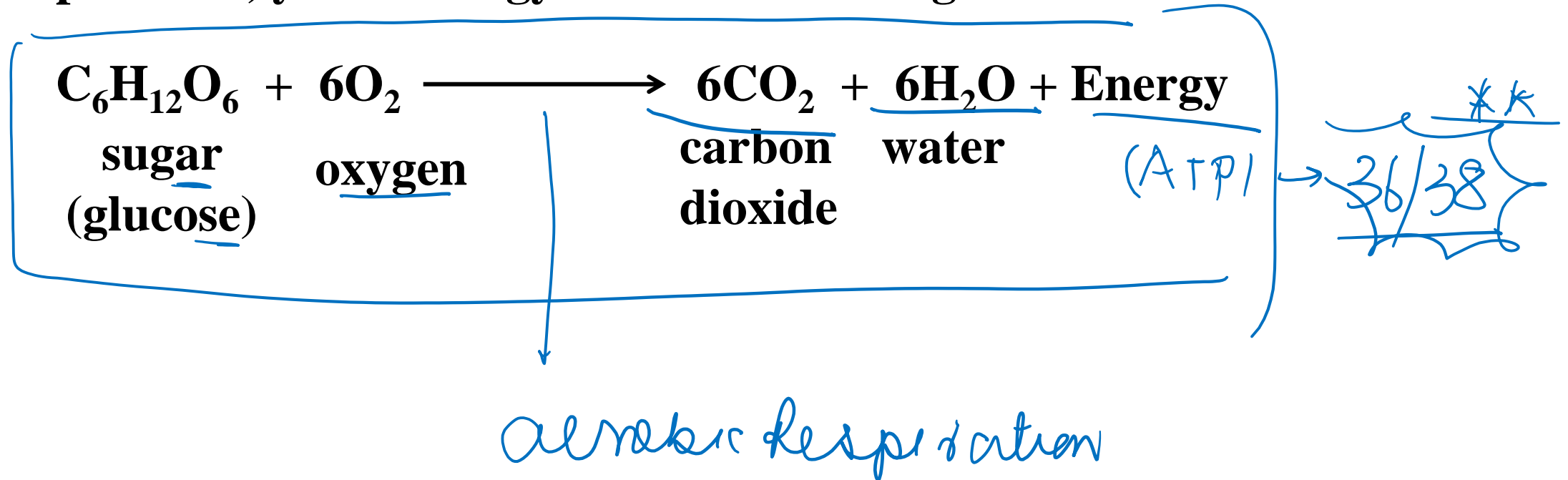
❖ **There are several reasons why plants can get along without respiratory organs.**

- (i) Each plant part takes care of its own gas-exchange needs.**
- (ii) Plants do not present great demands for gas exchange.**

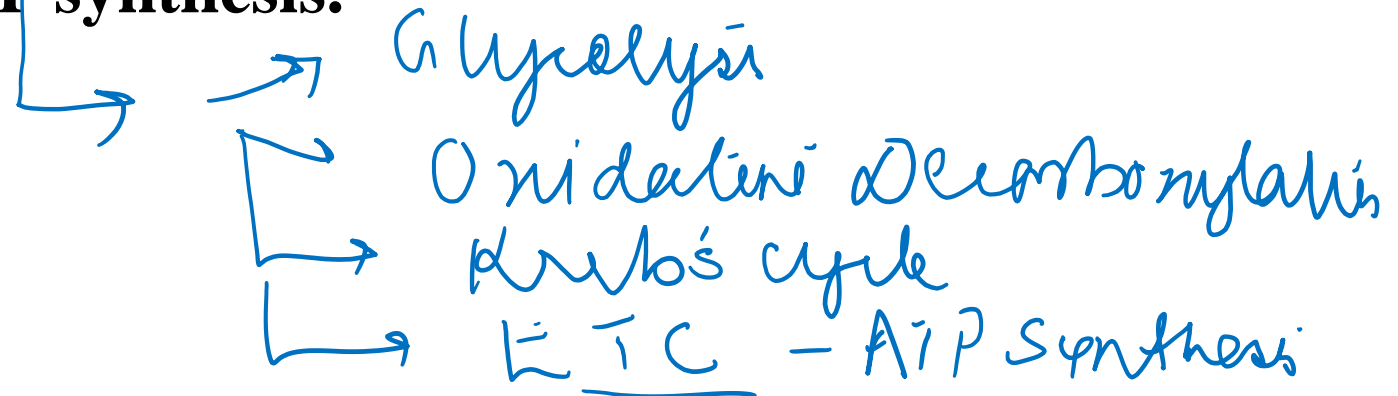
- **Roots, stems and leaves respire at rates far lower than animals.**

- **Only during photosynthesis large volumes of gases are exchanged and, each leaf is well adapted to take care of its own needs during these periods.**
- **When cells photosynthesise, availability of oxygen is not a problem in these cells since, oxygen is release within the cell.**

- ❖ The complete combustion of glucose, which produces CO_2 and H_2O as end products, yields energy most of which is given out as heat.



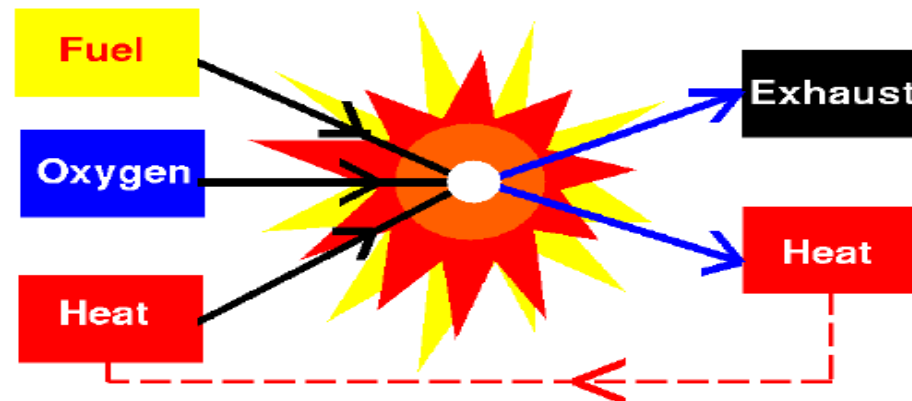
- ❖ If this energy is to be useful to the cell, it should be able to utilise it to synthesize other molecules that the cell requires.
- ❖ The strategy that the plant cell uses is to catabolize the glucose molecule in such a way that not all the liberated energy goes out as heat.
- ❖ The key is to oxidize glucose not in one step but in several small steps enabling some steps to be just large enough such that the energy released can be coupled to ATP synthesis.



- ❖ During the process of respiration, oxygen is utilised, and CO_2 and H_2O , energy are released as products.
- ❖ The combustion reaction requires oxygen.

What is combustion?

The process of burning something or rapid chemical combinations of a substance with oxygen involving production of heat and light.



Several organisms adapted to anaerobic conditions: some are facultative and others are Obligate anaerobes.

Microorganism that
and grow
molecul

An organism that
can grow only in the
complete absence of
molecular oxygen.

→ Obligate Anaerobe

→ QNEET

Facultative anaerobe → Can also
survive in the presence of O_2 .

- ❖ Pyruvic acid is then the key product of glycolysis.
- ❖ There are three major ways in which different cells handle pyruvic acid produced by glycolysis.
- ❖ These are lactic acid fermentation, alcoholic fermentation and aerobic respiration.

What is the
metabolic fate
of pyruvate?

This depends on the
cellular need.

FERMENTATION

anaerobic Resp in microorganisms

- ❖ Fermentation takes place under anaerobic conditions in many prokaryotes and unicellular eukaryotes.
- ❖ For the complete oxidation of glucose to CO_2 and H_2O , however, organisms adopt Krebs' cycle which is also called as aerobic respiration.

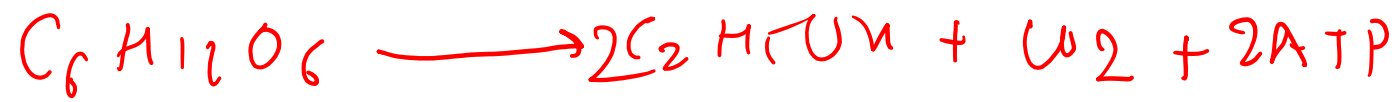
This required
 O_2 supply.

Alcoholic Fermentation

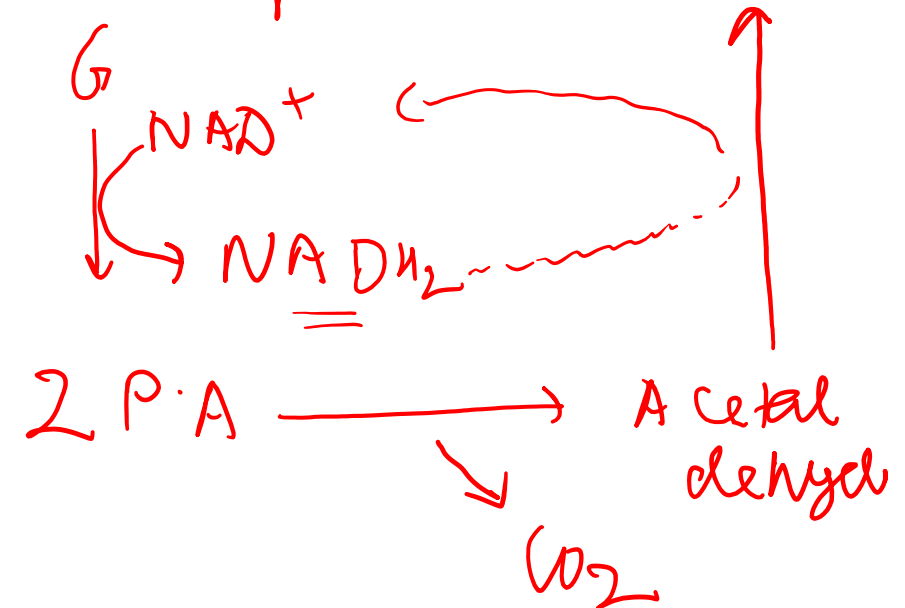
- ❖ In fermentation, say by yeast, the incomplete oxidation of glucose is achieved under anaerobic conditions by set of reactions where pyruvic acid is converted to CO₂ and ethanol.
- ❖ The enzymes, pyruvic acid ^{*}decarboxylase and ^{*}alcohol dehydrogenase catalyse these reactions.
↳ remove CO₂
- ❖ In some bacteria lactic acid formed from pyruvic acid in presence of lactate dehydrogenase. [No decarboxylation] → ** QNUT
- ❖ The reducing agent is NADH+H⁺.

❖ In both fermentations ^{alcohol or lactic acid} not much energy is released; less than *seven percent of the energy in glucose is released and not all of it is trapped as high energy bonds of ATP.

❖ When one molecule of glucose is fermented to alcohol or lactic acid only two molecules of ATP are formed.



→ only 2 ATP is formed. L-manol

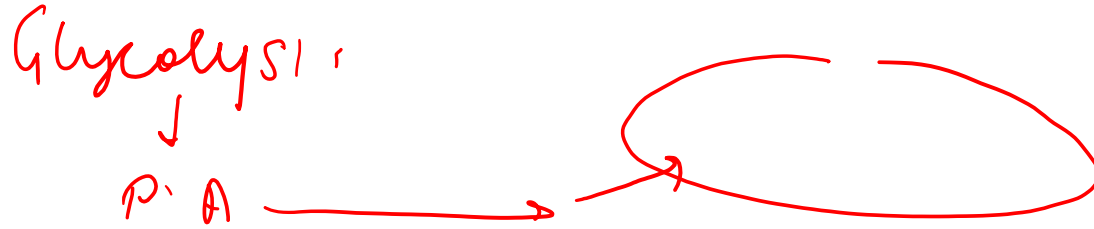


Representation:

NET 2

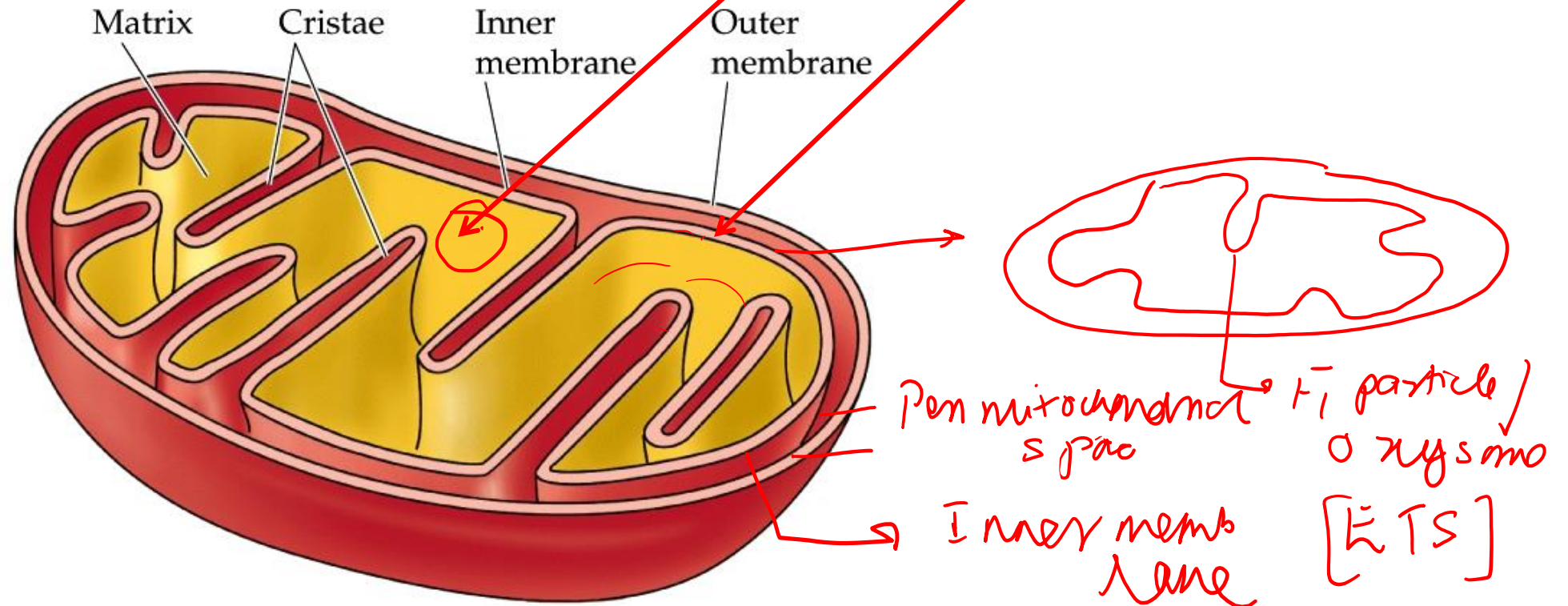
- ❖ What then is the process that yeast carry out complete oxidation of glucose when the concentration of alcohol reaches about 13%? to synthesise a large number of cellular metabolism?
- ❖ In eukaryotes these steps take place within the mitochondria and this required O_2 .
- ❖ Aerobic respiration is the process that leads to a complete oxidation of organic substances in the presence of oxygen, and release CO_2 , water and a large amount of energy present in the substrate.
- ❖ This type of respiration is most common in higher organisms.

Aerobic respiration



- Aerobic respiration takes place within mitochondria.
- Pyruvic acid is transported from the cytoplasm to the mitochondria.
- The important events in aerobic respiration are :
 1. The complete oxidation of pyruvic acid by the stepwise removal of all the hydrogen atoms, leaving three molecules of CO_2 . *Kreb's cycle*
 2. The passing of the electrons removed as a part of hydrogen atoms to molecular O_2 with simultaneous synthesis of ATP. *↳ ETS*

- The 1st event takes place in the matrix of the mitochondria. *Kreb's cycle*
- The 2nd event takes place on the inner membrane of the mitochondria.



GLYCOLYSIS

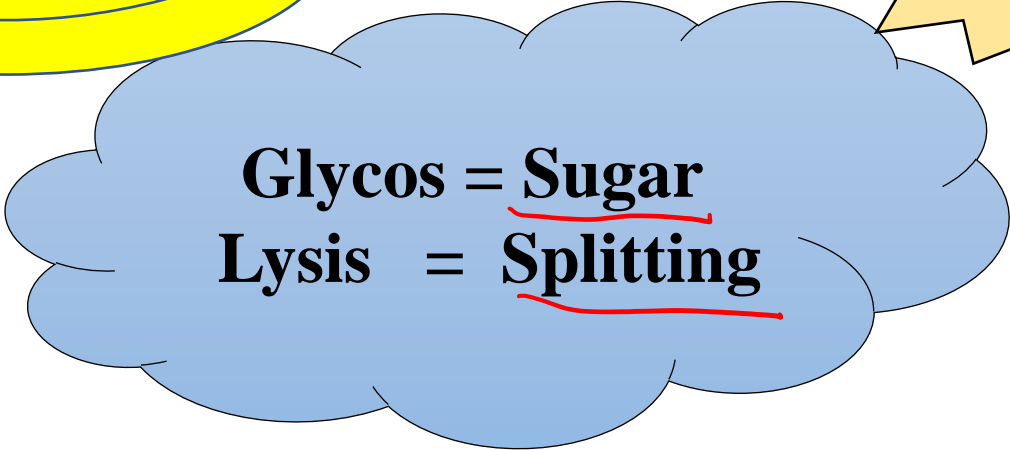
GLYCOLYSIS



Means



**In Greek
words**

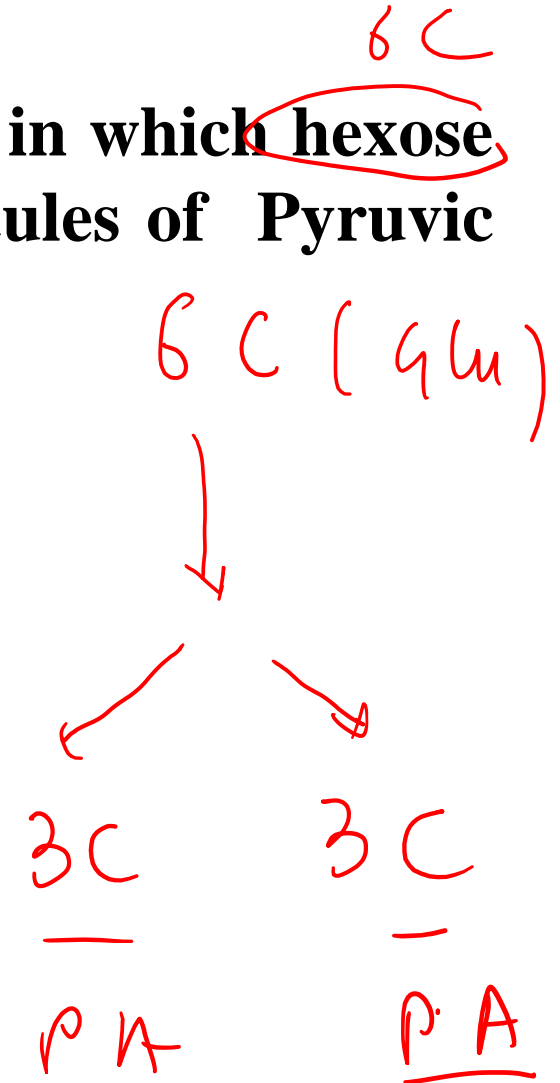


Glycos = Sugar
Lysis = Splitting

GLYCOLYSIS

- ❖ Glycolysis is the first stage of respiratory metabolism in which hexose sugar undergoes partial oxidation to form two molecules of Pyruvic acid (3 carbon compound).

Glycolysis is the common step for both aerobic and anaerobic respiration and occurs in the cytosol.



❖ The sequence of reactions in glycolysis were traced out by German scientists Embden, Meyerhoff and Paranas, it is known after them as “**EMP pathway**”.



Embden



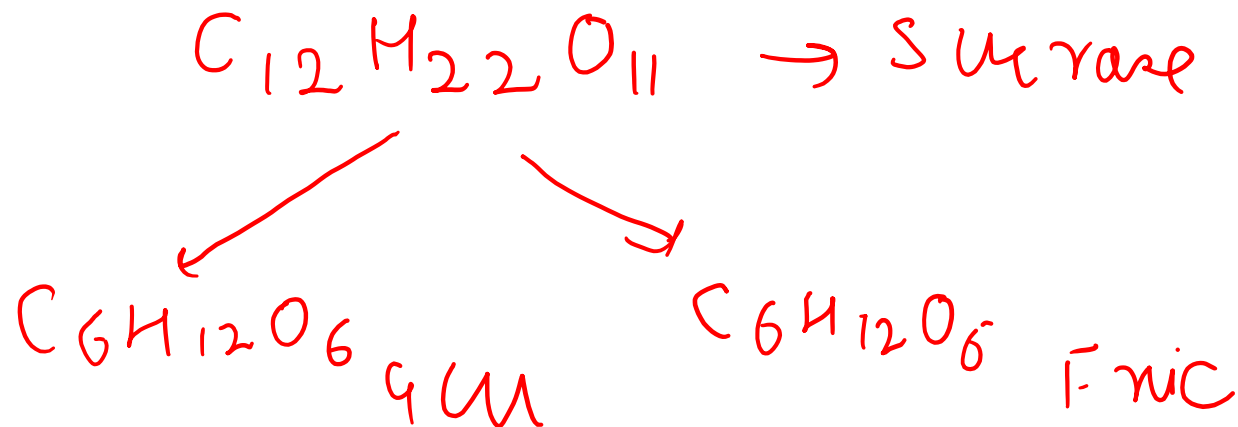
Meyerhoff



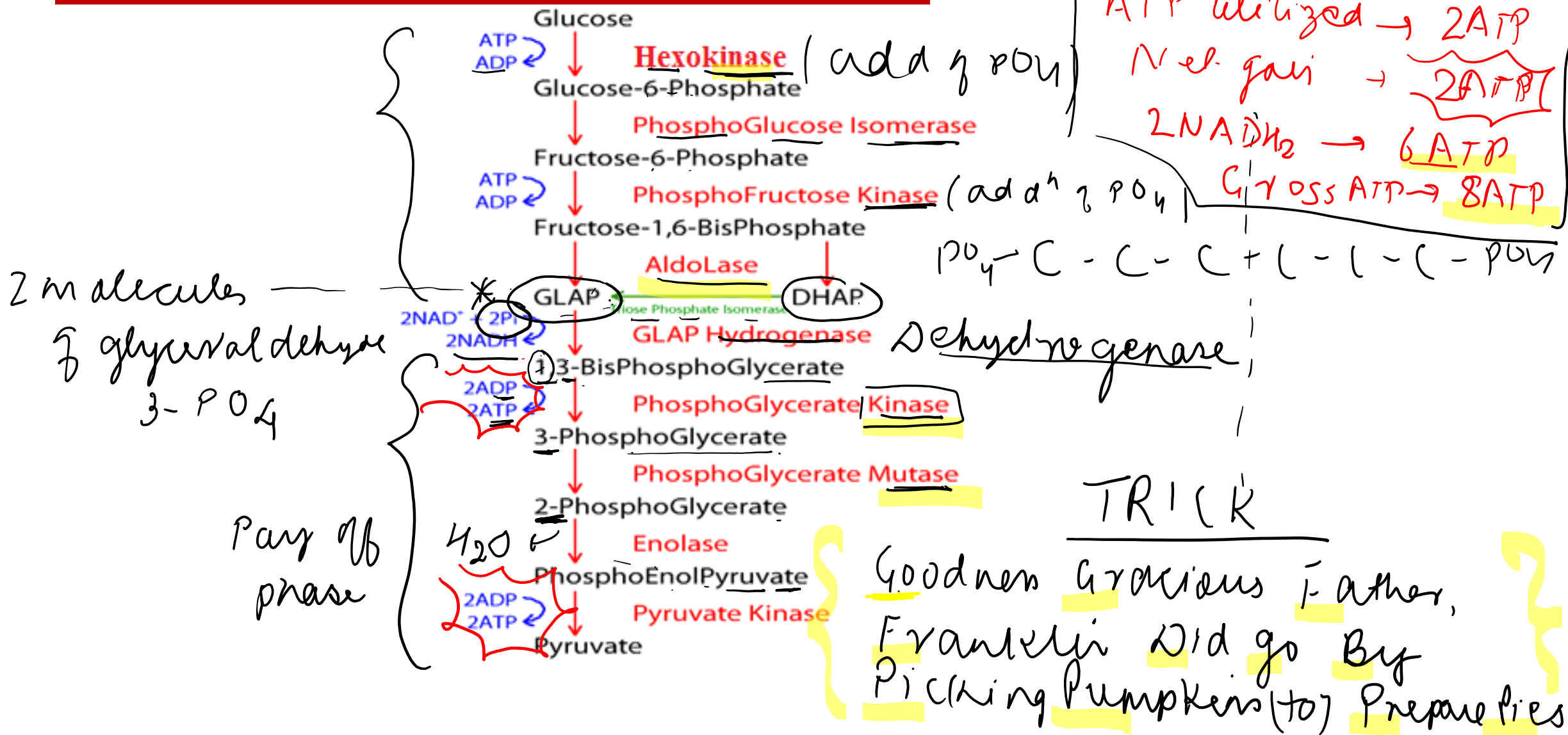
Paranas

Dissaccharide

- ❖ Sucrose is converted into glucose and fructose by the enzyme invertase, and these two monosaccharides readily enter the glycolytic pathway.
- ❖ Glucose and fructose are phosphorylated to give rise to glucose-6-phosphate in presence of **hexokinase**.
- ❖ Glucose-6-phosphate is isomerised to fructose-6-phosphate, in presence of hexose phosphate isomerase.



Reactions of the Glycolytic sequence



- ❖ Fructose-6-phosphate reacts with ATP to form fructose 1, 6-bisphosphate, enzyme is **phosphofructokinase**.
- ❖ The fructose 1, 6-bisphosphate is split into DHAP and PGAL.
- ❖ This reaction is catalyzed by **aldolase**.
- ❖ DHAP is also isomerized to PGAL, this reaction is catalyzed by **triose phosphate isomerase**.
- ❖ PGAL is oxidised and with inorganic phosphate get converted into BPGA(1, 3 bisphosphoglycerate).

- ❖ This oxidation gives 2NADH_2 .
- ❖ This reaction is catalyzed by **glyceraldehyde-3-phosphate dehydrogenase**.
- ❖ The conversion of BPGA to PGA is also an energy yielding process.
- ❖ This energy is trapped in the form of ATP.
- ❖ Another ATP is synthesized during the conversion of PEP to pyruvic acid.

- ❖ Pyruvic acid is then the key product of glycolysis.
- ❖ There are three major ways that different cells handle pyruvic acid produced by glycolysis.
- ❖ These are lactic acid fermentation and aerobic respiration.

What is the metabolic fate of pyruvate?

This depends on the cellular need.

AIR

LCF

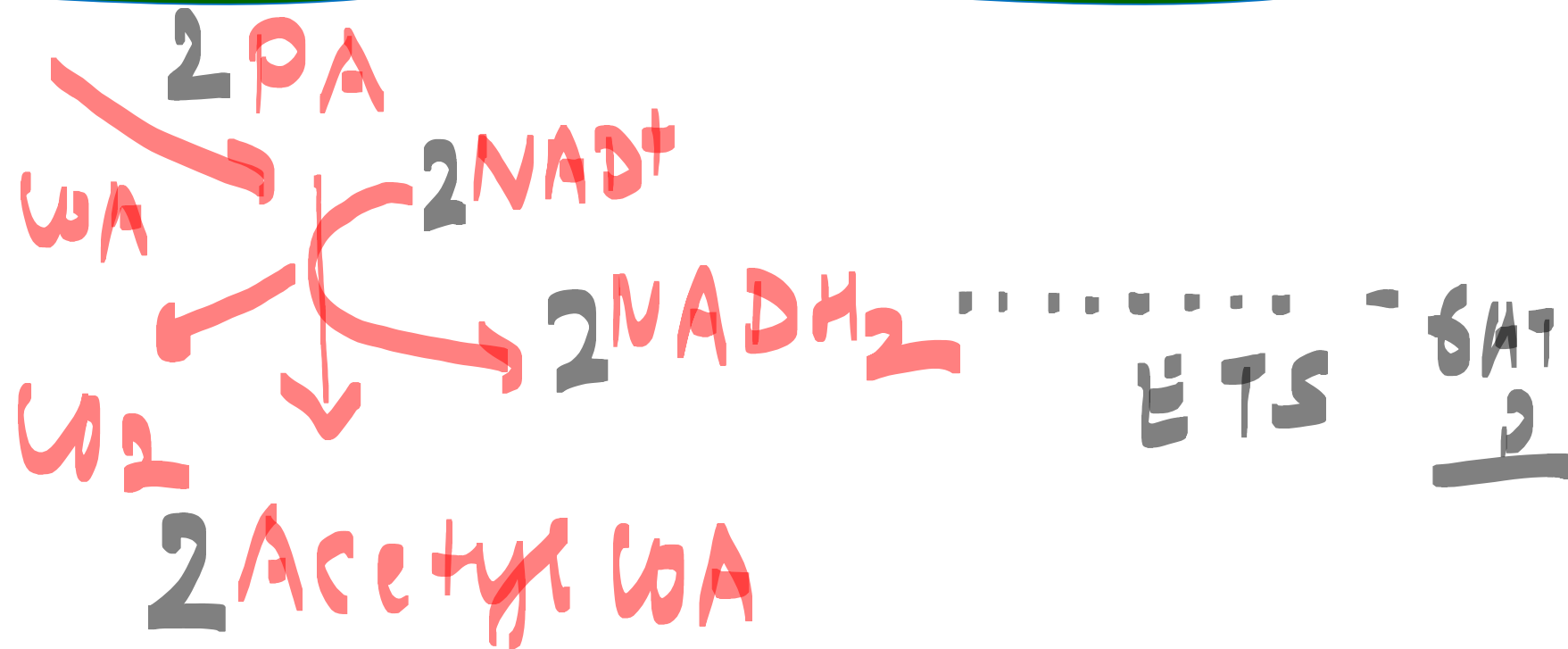
AP

- ❖ Fermentation takes place under anaerobic conditions in many prokaryotes and unicellular eukaryotes.
- ❖ For the complete oxidation of glucose to CO_2 and H_2O , however, organisms adopt Krebs' cycle which is also called as aerobic respiration.



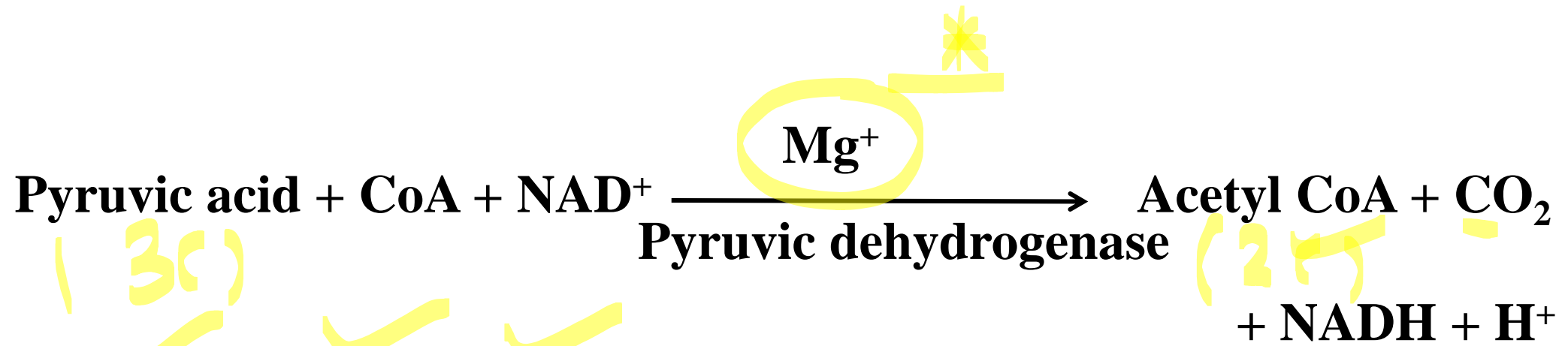
**This required
 O_2 supply.**

Oxidative decarboxylation of Pyruvic acid



Occurs in the Mitochondrial matrix.

- ❖ **Pyruvate which is formed by the glycolytic catabolism of carbohydrates in the cytosol, after entering mitochondrial matrix, undergoes oxidative decarboxylation by a complex set of reactions catalysed by pyruvic dehydrogenase.**
- ❖ **The reactions catalysed by pyruvic dehydrogenase require the participation of several coenzymes, including NAD^+ and Coenzyme A.**



- ❖ During this process, two molecules of NADH are produced from the metabolism of two molecules of pyruvic acid (produced from one glucose molecule during glycolysis).
- ❖ The acetyl CoA then enters a cyclic pathway, TCA cycle, more commonly called **Krebs-cycle**.

AEROBIC RESPIRATION- TRICARBOXYLIC ACID CYCLE

Citric Acid cycle or Tricarboxylic Acid cycle or Krebs Cycle

- ❖ In 1937 English Biochemist Sir Hans Krebs postulated the cycle of reactions which describe the oxidation of pyruvate to CO_2 and came to be known as Citric acid cycle.
- ❖ In 1953, Krebs was awarded Nobel prize, and the cycle is frequently referred to as Krebs cycle in his honour.



Hans Krebs, 1900–1981

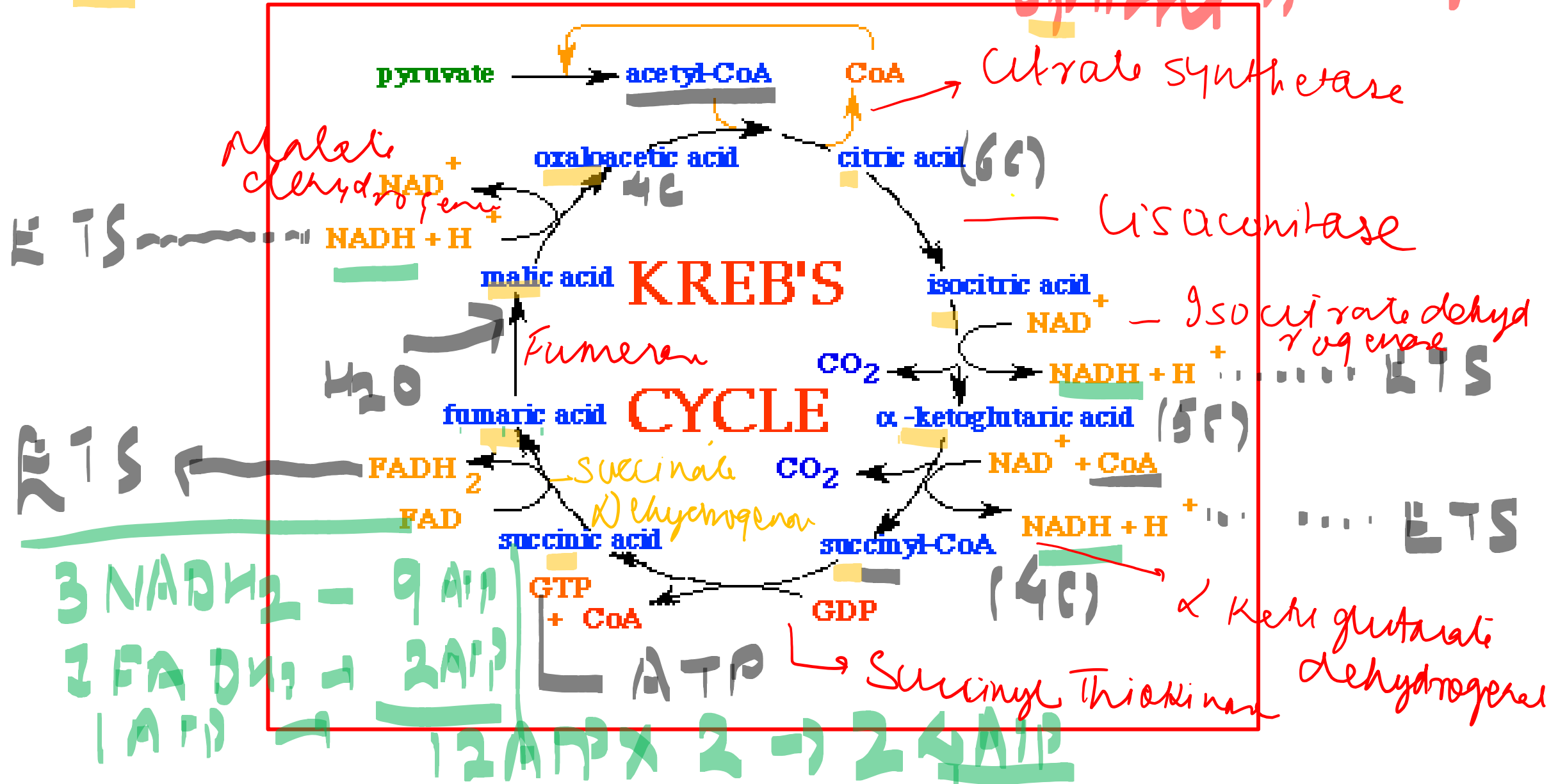
Citric Acid cycle or Tricarboxylic Acid cycle or Krebs Cycle

- ❖ **The TCA cycle starts with the condensation of acetyl group with OAA and water to yield citric acid.**
- ❖ **This reaction is catalysed by citrate synthase and a molecule of CoA is released.**
- ❖ **Citrate is then isomerised to isocitrate.**
- ❖ **It is followed by two successive steps of decarboxylation, leading to the formation of α -Ketoglutaric acid and then succinyl-CoA.**

- ❖ **During the conversion of Succinyl CoA to Succinic acid a molecule of GTP is synthesized. This is a substrate level phosphorylation.**
- ❖ **In a coupled reaction GTP is converted to GDP with the simultaneous synthesis of ATP from ADP.**
- ❖ **At 3 points NAD^+ is reduced to $\text{NADH} + \text{H}^+$ and at one point FAD^+ is reduced to FADH_2 .**
- ❖ **The continued oxidation of acetyl CoA via the TCA cycle requires the continued replenishment of oxaloacetic acid, the first member of the cycle.**

- ❖ In addition it also requires regeneration of NAD^+ and FAD^+ from NADH and FADH_2 respectively.
- ❖ The equations for this phase of respiration may be written as follows:

"I know some simple formula making option"



Sequence of Krebs cycle is :

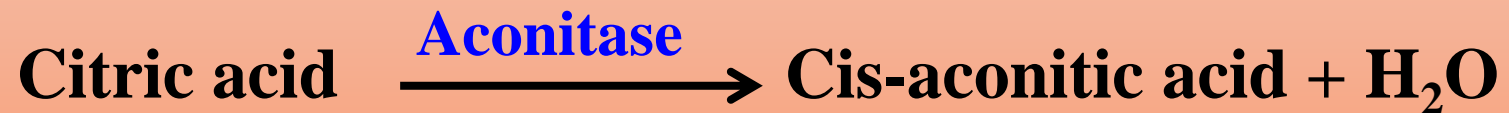
1.

CONDENSATION



2.

DEHYDRATION



Sequence of Krebs cycle is :

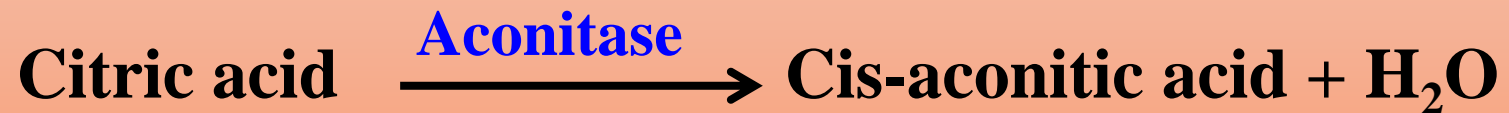
1.

CONDENSATION



2.

DEHYDRATION



3.

HYDRATION



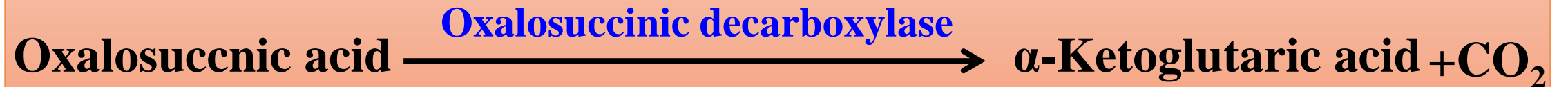
4.

OXIDATION -I



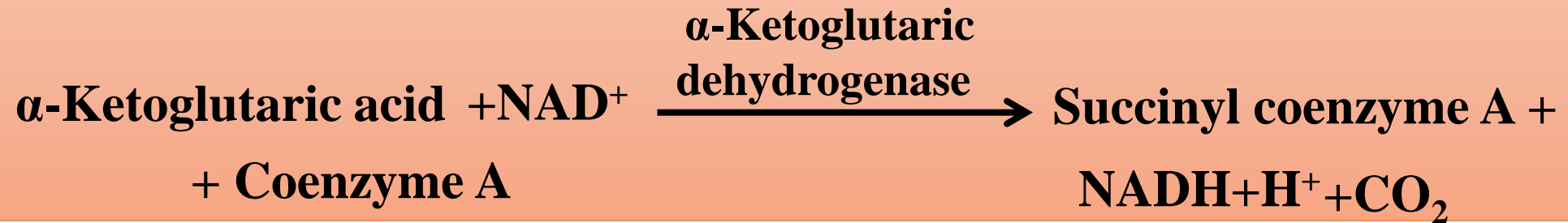
5.

DECARBOXYLATION



6.

OXIDATIVE DECARBOXYLATION(OXIDATION II)



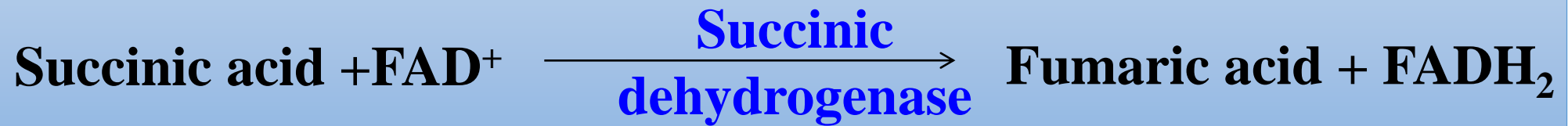
7.

CLEAVAGE



8.

OXIDATION III



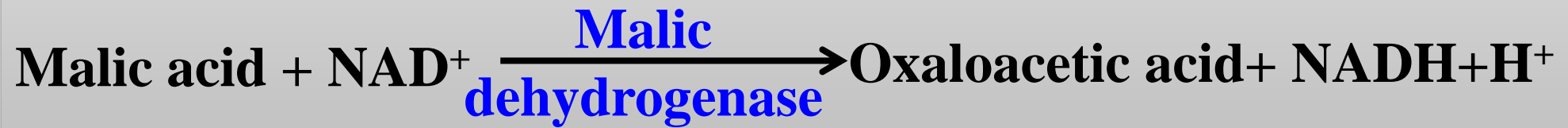
9.

HYDRATION



10.

OXIDATION IV



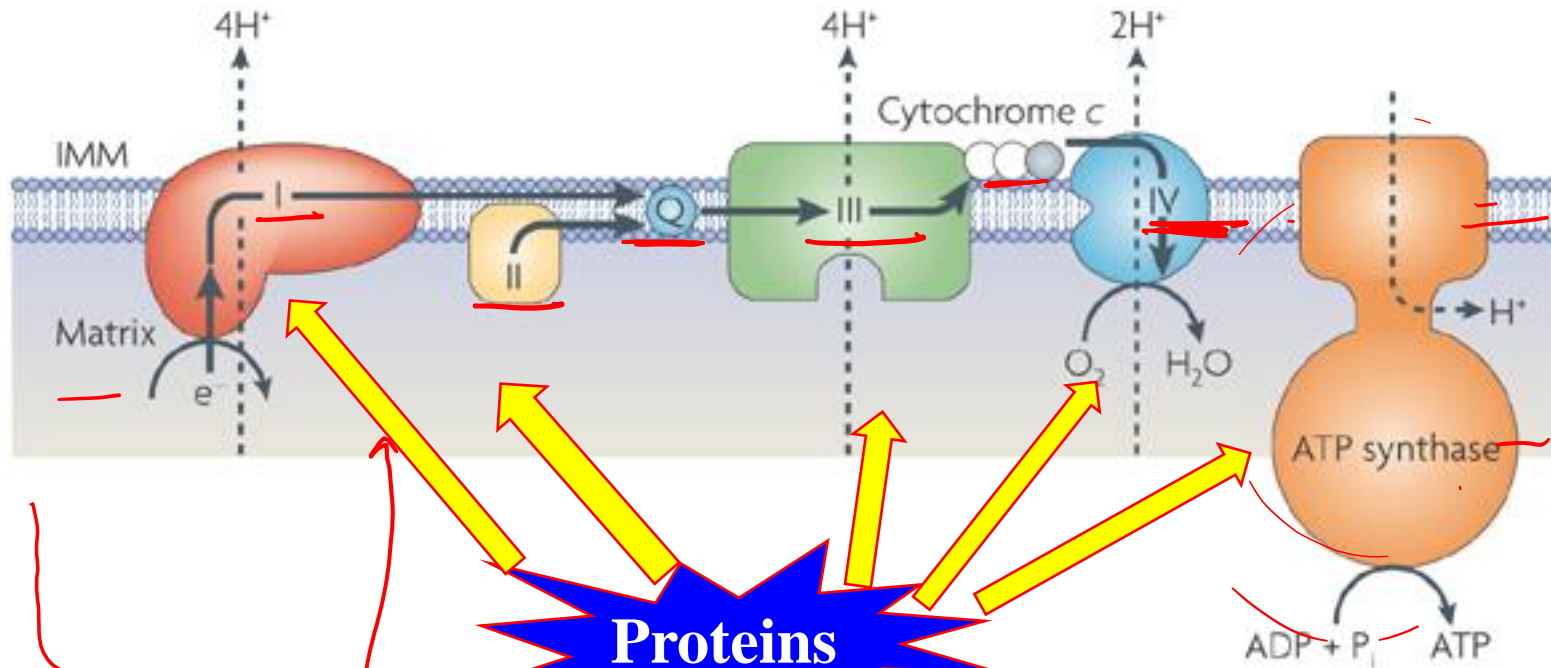
Summary equation for the oxidation of pyruvic acid in mitochondrion is:



Final Rk's Krebs
cycho

ELECTRON TRANSPORT SYSTEM AND OXIDATIVE PHOSPHORYLATION

Electron Transport System and Oxidative Phosphorylation



ATP in
the
ETS

via
NADH
FADH₂

Proteins

❖ Series of proteins in the mitochondria helps transfer electrons (e^-) from NADH to Oxygen.

- inner membrane

Integral proteins

Penm
cha
space
outer
mem

Most energy from Redox reactions

- ❖ Electrons during metabolic reactions sent to NAD and FAD.

Electron transport system

No. of NADH/FADH₂
produced at various
levels of aerobic
respiration.

- ❖ 5 Multiprotein complexes
- ❖ 2 mobile Carriers: Ubiquinone and Ubiquinol

Chromosome 'C'.

** QINLET

Complex I (NADH – dehydrogenase)

❖ It transfers electrons from NADH to Ubiquinone.

Ubiquinone
located within the
inner membrane.

Ubiquinone is the
mobile carrier between
complexes I and III and
also between II and III.

Complex II (Succinate dehydrogenase)

- ❖ **Transfers electrons from Succinate to Ubiquinone (FADH₂ to Ubiquinone).**

Complex III (Cytochrome “c” reductase/cytochrome bc₁ complex)

- ❖ **Catalyse the reduction of Cytochrome c by the transfer of electrons from Ubiquinol.**

Cytochrome “c” is a small protein attached to the outer surface of the inner membrane.

Complex IV (Cytochrome "c" oxidase)

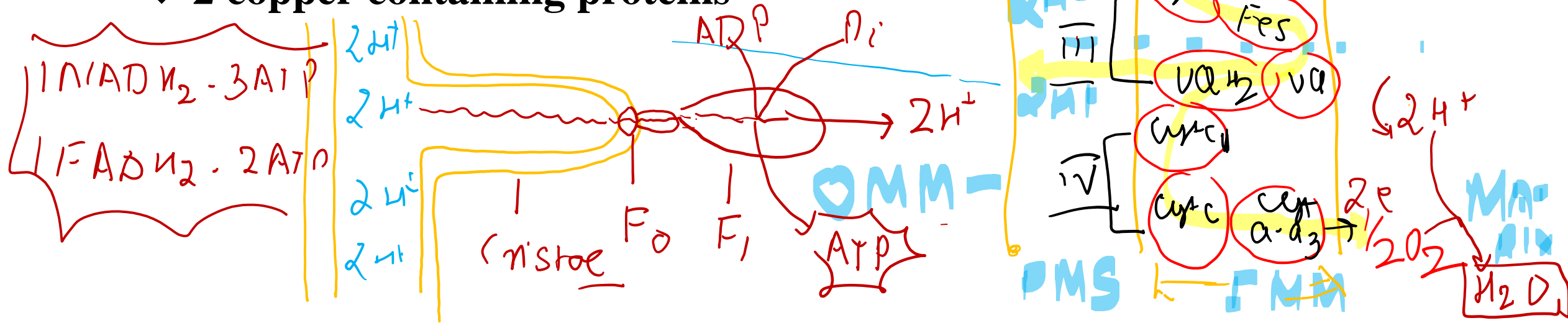
❖ Transfers electrons from reduced cytochrome 'c' to molecular oxygen.

It Contains four units :

❖ Cytochrome 'a'

❖ Cytochrome 'a₃'

❖ 2 copper containing proteins



Respiratory Balance sheet:

- ❖ **It is possible to make calculations of the net gain of ATP for every glucose molecule oxidized; but in reality this can remain only theoretical exercise.**
- **There is a sequential, orderly pathway functioning, with one substrate forming the next and with glycolysis, TCA cycle and ETS pathway following one after another.**
- **The NADH synthesized in glycolysis is transferred into the mitochondria and undergoes oxidative phosphorylation.**

- **None of the intermediates in the pathway is utilized to synthesize any other compound.**

- **But these reactions are not coupled in a living system.**
- **All the reactions take place one after the other.**
- **Substrates are entering the pathway at any of the intermediary stages.**
- **Products are withdrawn from it as and when necessary and when needed.**

- **Enzymatic rates are controlled by multiple means.**
- ❖ **Yet, it is useful to do this exercise to appreciate the beauty and efficiency of the living system in extracting and storing energy.**
- ❖ **Hence, there can be a net gain of 36 ATP molecules during aerobic respiration of one molecule of glucose.**

❖ Now, let us compare fermentation and aerobic respiration.

<u>Fermentation</u>	<u>Aerobic respiration</u>
i) <u>Partial oxidation of glucose</u> takes place	<u>Complete oxidation of glucose</u> takes place
ii) Net gain of ATP is <u>2</u>	36 molecules (Net) of ATP is <i>vial glycerol</i> formed
iii) NADH is oxidised to NAD slowly	<i>shunt</i> It is <u>vigorous</u>

Balance sheet of ATP production in aerobic oxidation of glucose

1) Glycolysis

i) ATP produced by substrate level phosphorylation

❖ 1,3- biphosphoglyceric acid to 3- phosphoglyceric acid $2 \times 1 = 2 \text{ ATP}$

❖ Phosphoenol pyruvic acid to pyruvic acid

$2 \times 1 = 2 \text{ ATP}$

4 ATP



ATP Consumed

- ❖ Glucose to Glucose – 6 - phosphate - 1 ATP
- ❖ Fructose – 6- phosphate to Fructose , 1,6- biphosphate - 1 ATP

+ 4 ATP

Net gain of ATP (4 - 2)

+2 ATP

* Imp NET

2. ATP from NADH generated in glycolysis:

- ❖ G-3-P to BPGA (2NADH, each worth 2 ATP) $2 \times 2 \text{ ATP} = 4 \text{ ATP}$
- ❖ ATP gain from glycolysis in the presence of O_2

6 ATP ----- (a)

↳ NADH_2 via ETS

II. Oxidative decarboxylation of Pyruvic acid

Pyruvic acid to acetyl CoA (2NADH, each worth 3ATP)

From 2NADH ($2 \times 3 \text{ ATP}$) = 6ATP----- (b)


↳ via ETS

III Krebs (citric acid) cycle

i. ATP produced by substrate level phosphorylation

Succinyl Co A to Succinic acid $2 \times 1 \text{ ATP} = 2 \text{ ATP}$

ii. From NADH

- ❖ Isocitric acid to Oxalosuccinic acid $2 \times 3\text{ATP} = 6 \text{ ATP}$
 - ❖ α -Ketoglutaric acid to Succinyl CoA $2 \times 3\text{ATP} = 6 \text{ ATP}$
 - ❖ Malic acid to Oxaloacetic acid $2 \times 3\text{ATP} = 6 \text{ ATP}$
- 

iii. From FADH_2

Succinic acid to Fumaric acid

$$2 \times 2\text{ATP} = 4 \text{ ATP}$$

ATP value of Krebs cycle(i+ii+iii)

24 ATP

----- (c)

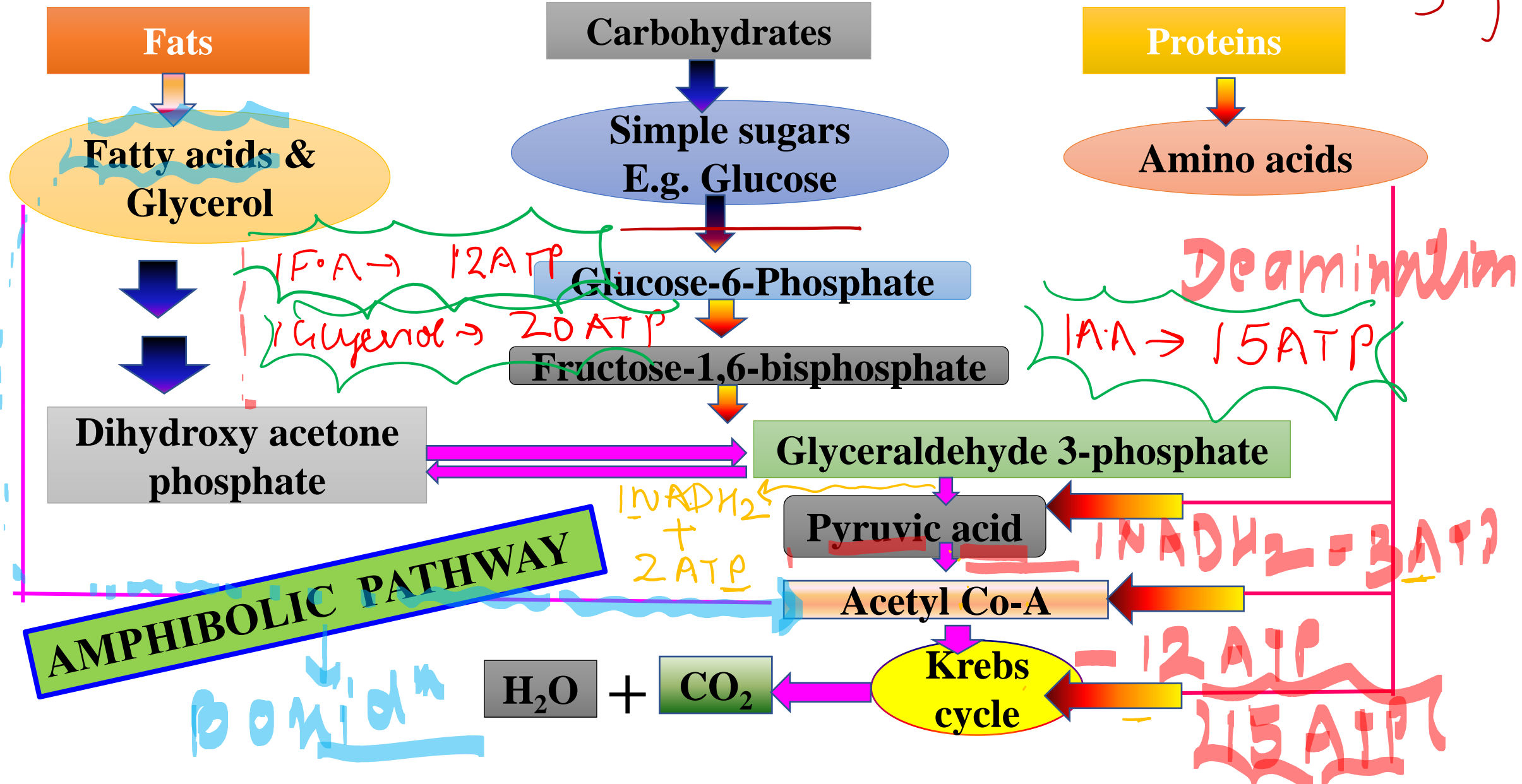
Net gain of ATP in aerobic respiration
per mole of glucose(a+b+c)

36 ATP

~~9 from~~ Krebs's cycle

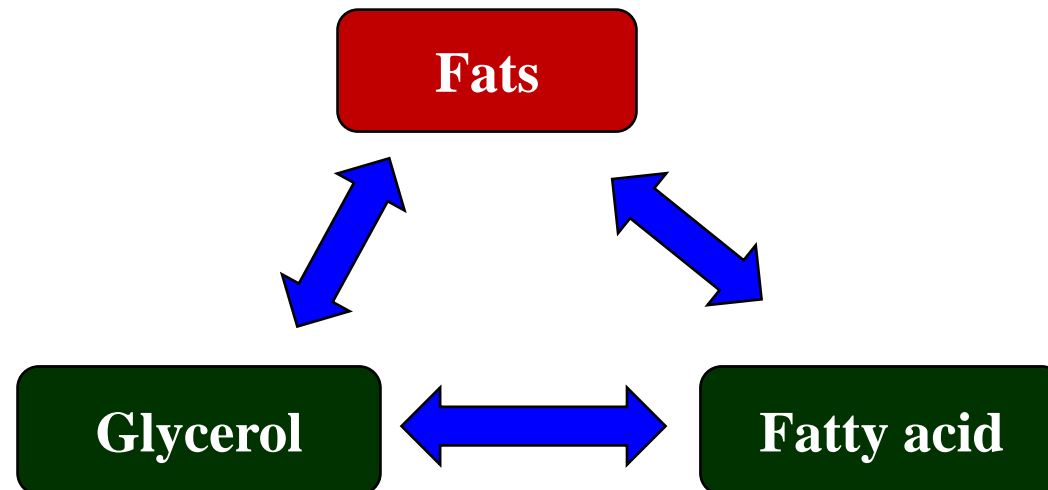
AMPHIBOLIC PATHWAY

Carbohydrat \rightarrow Fat \rightarrow Protein (has to be metabolized)



- ❖ Glucose is the most favoured substrate for respiration.
- ❖ All carbohydrates are converted into glucose before they are used for respiration.
- ❖ Entry of different substrates in respiratory pathway:

Fats:



- ❖ If fatty acids to be respired first it would be degraded to acetyl Co-A.

Fatty acids → Acetyl Co-A

- ❖ Glycerol enter the pathway after being converted to PGAL.

Glycerol → PGAL

- ❖ Proteins degraded by Proteases and the individual amino acids (after deamination).

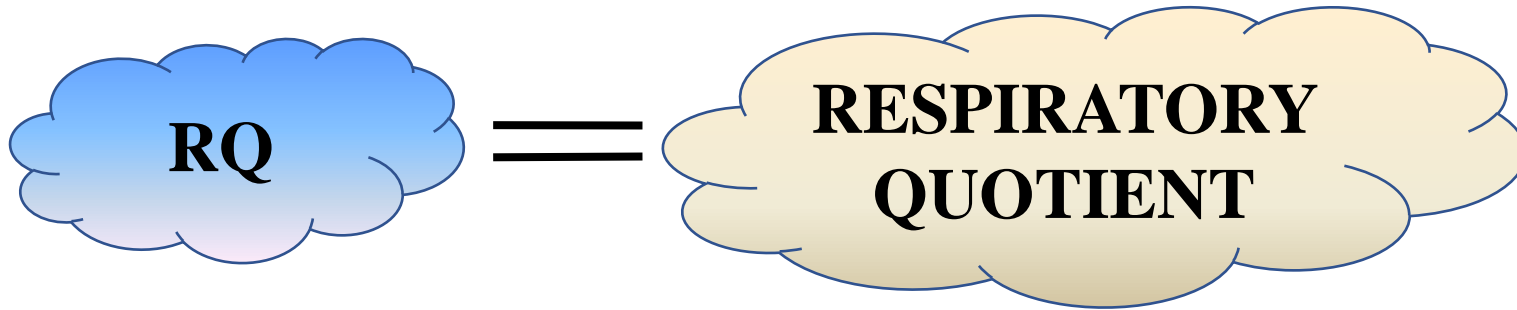
} crisp

- ❖ Respiration involves the breakdown of substrates, the respiratory process has traditionally been considered as **Catabolic process** and respiratory pathway as a **catabolic pathway**.
- ❖ Hence, fatty acids would be broken down to Acetyl CoA before entering the respiratory pathway when it is used as a substrate.
- ❖ But when the organism needs to synthesize fatty acids, Acetyl CoA would be withdrawn from the respiratory pathway for it.
- ❖ Hence, respiratory pathway comes into the picture both during the breakdown and synthesis of fatty acids, similarly breakdown and synthesis of proteins too.

❖ The breaking down process within the living organisms constitute **Catabolism**, while the synthesis is **Anabolism**.

The respiratory pathway is involved in both anabolism and catabolism.

The respiratory pathway is considered as Amphibolic pathway rather than catabolic process.



- ❖ “The ratio between the volume of **Carbondioxide given out** and **oxygen taken** in respiration.

$$\text{RQ} = \frac{\text{Volume of CO}_2 \text{ evolved}}{\text{Volume of O}_2 \text{ consumed}}$$

→ provides idea of The substrate being used.

❖ **RQ is an index of type of substrate being respired.**

**Different respiratory
substrates yield
different RQ values.**

Respiratory substrates

Carbohydrates

Proteins

Fats

Organic acids

Carbohydrates:

RQ value "1"

❖ When carbohydrates are used as substrate and are completely oxidised, the RQ is 1.

e.g.

Glucose

CO₂ evolved = O₂
consumed

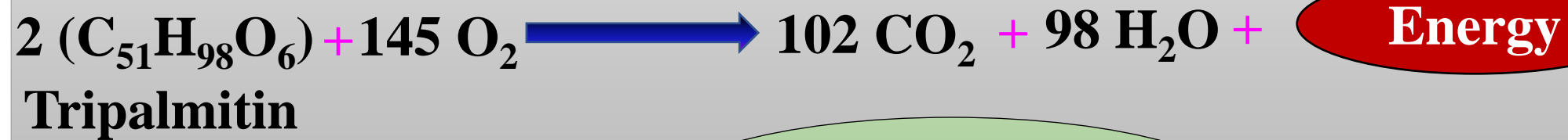
$$RQ = \frac{6CO_2}{6O_2} = 1$$



$$R.Q = \frac{6}{6} = 1$$

Fats:

- ❖ When fats are used in respiration, the RQ is less than 1. Calculations for a fatty acid, tripalmitin, if used as a substrate is shown below:



RQ value is
"0.7"

$$\text{RQ} = \frac{102 \text{CO}_2}{145 \text{O}_2}$$

↳ Fat & protein
R.Q is less than
one

Protiens :

- ❖ When proteins are respiratory substrates the ratio would be about 0.9.
- ❖ What is important to recognise is that in living organisms respiratory substrates have RQ value more than 1.

Pure proteins or fats are never used as respiratory substrates.

For anaerobic respiration
 $C_6H_{12}O_6 \rightarrow 2C_2H_5OH + 2CO_2$

$$R.Q. = \frac{2}{0} = \infty$$

Succulent plants Malic Acid
 $C_6H_{12}O_6 + 3O_2 \rightarrow C_4H_4O_5 + H_2O$
 $\frac{0}{3O_2} = 0$